

Oxyanion Nomenclature

(Naming Oxyanions)

By Dr. Shawn P. Shields



This work is licensed by Shawn P. Shields-Maxwell under a [Creative Commons Attribution-NonCommercial-ShareAlike 4.0 International License](https://creativecommons.org/licenses/by-nc-sa/4.0/).

Oxyanions

- Oxyanions are polyatomic ions involving one or more oxygen atoms plus another nonmetal atom.
- Oxyanions commonly have negative charges of -1 , -2 , and -3 .
- Examples include:
 - CO_3^{2-} "Carbonate"
 - NO_2^- "Nitrite"
 - SO_4^{2-} "Sulfate"

General Rules for Naming Oxyanions

- Oxyanions are named according to the steps given below.
 - 1) Write the root of the name of the non-oxygen element
 - 2) Add the *-ate* or *-ite* suffix to the end of the name, according to the rule on the next slide.
- Examples:
 - NO_3^- Nitrogen is the non-oxygen element. Remove the *-ogen* and add *-ate* "nitrate"
 - SO_3^{2-} Sulfur is the non-oxygen element. Remove *-ur* and add *-ite* "sulfite"

Oxyanion Nomenclature Patterns

- Often, there will be more than one oxyanion involving a given element.
- The difference is the number of oxygen atoms bonded.
- For example:
 - NO_2^- (nitrite) and NO_3^- (nitrate)
 - SO_3^{2-} (sulfite) and SO_4^{2-} (sulfate)
- Notice that the charge on the polyatomic ion is the same for both anions in the pair.

Oxyanion Nomenclature (Naming)

- Compare the sets of oxyanions below. Notice the pattern in their names.
 - NO_2^- (nitrite) and NO_3^- (nitrate)
 - SO_3^{2-} (sulfite) and SO_4^{2-} (sulfate)
 - ClO_2^- (chlorite) and ClO_3^- (chlorate)
- The one with fewer oxygens ends in *-ite*.
- The one with more oxygens ends in *-ate*.

Naming Oxyanion Series

- Let's go back to the oxyanions involving chlorine below:

- ClO_2^- (chlorite) and ClO_3^- (chlorate)

- It turns out that two more anions exist

ClO^- (hypochlorite) vs. ClO_4^- (perchlorate)

- The one with fewer oxygens than the -ite anion has a prefix *hypo-* added at the beginning of the name.
- The one with more oxygens than the -ate anion has a prefix *per-* added at the beginning of the name.

Mini Quiz

- Name the oxyanions given below:



Mini Quiz Solutions

- Name the oxyanions given below:



Mini Quiz

- Name the series of oxyanions given below:



Mini Quiz Solutions

- Name the series of oxyanions given below:

- IO^- (hypoiodite)

- IO_2^- (iodite)

- IO_3^- (iodate)

- IO_4^- (periodate)